# Preparation of 4'-Amino Hexopyranose Nucleosides from Keto Nucleosides. An Approach to Amino Nucleoside Antibiotics 

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#### Abstract

Amino hexopyranose nucleosides were prepared by treatment of the keto group with hydroxylamine hydrochloride followed by acetylation of the resulting oxime. Among other reagents, sodium borohydride-nickel chloride or sodium borohydride-molybdenum( v ) oxide were found to be excellent reductive systems for the facile synthesis of amino hexopyranose nucleosides.


The $4^{\prime}$-amino hexopyranose skeleton is found in several biologically active naturally occurring nucleosides including gougerotin and blasticidin ${ }^{1.2}$ which are of great interest due to their pronounced antibiotic and antitumoral activities. Synthetic approaches to these nucleosides have been reported to occur either by condensation of amino sugars with heterocyclic bases ${ }^{3.4}$ or by direct modification of a nucleoside. ${ }^{5}$ In both cases the amino function was introduced via an azido group by a substitution-reduction sequence.


Blasticidin S

Gougerotin

Considerably less attention has been given to biomimetic synthesis from keto nucleosides as has been hypothesized by Seto et al. ${ }^{6}$ This strategy offers a straightforward approach avoiding side reactions which are often significant during azide group introduction.?

In connection with our continuing interest in keto hexopyranosyl nucleosides ${ }^{8}$ we have recently shown that this class of biologically active compounds ${ }^{9}$ are important building blocks for the preparation of a wide range of new nucleosidic structures such as fused lactone nucleosides ${ }^{10}$ and spiro epoxy derivatives. ${ }^{11}$ We have also demonstrated that hydride reduction of isopropylidene and unsaturated nucleosides proceeds with a high degree of stereo- and chemo-selectivity. ${ }^{8}$ These results suggested that hydride addition to imino or hydroximino nucleosides prepared from keto nucleosides could lead to amino hexopyranose nucleosides in a straightforward and stereoselective manner.
In this paper we describe the synthesis of $2^{\prime}$ - and $4^{\prime}$-amino
nucleosides from readily accessible 6-deoxy-L-hexopyranosyl isopropylidene keto nucleosides and 6-deoxy-L-hexopyranosyl unsaturated keto nucleosides. The choice of these nucleosides as models originates from the idea that 6-deoxy-2- and -4-L-amino nucleosides may be expected to offer new antineoplastic propertits since 6-deoxy-L-amino sugars constitute the carbohydrate part of a great number of naturally occurring biologically active glycosides. ${ }^{12}$

## Results and Discussion

The approach employed initially was based on the reaction of keto nucleosides with sodium cyanoborohydride in the presence of ammonium acetate. ${ }^{13}$ Reaction of 1-( $6^{\prime}$-deoxy- $3^{\prime}, 4^{\prime}-O$-iso-propylidene- $\beta$-L-lyxo-hexopyranosulosyl)thymine ${ }^{10}$ (1a) with this system provided mainly the talo-nucleoside (2). A suitable strategy for amino nucleoside preparation ${ }^{14,15}$ would involve conversion of the ketone into an oxime followed by reduction of the latter. Thus compound (1a) treated with hydroxylamine hydrochloride afforded the oxime nucleoside (3a). The product isolated in $70 \%$ yield consisted predominantly of the $Z$-isomer ${ }^{9}$ (2:1) as established from ${ }^{1} \mathrm{H}$ n.m.r. spectroscopy $\left\{\delta_{\mathrm{H}} 6.7[0.66 \mathrm{H}\right.$, $\left.\left.\mathrm{s}, 1^{\prime}-\mathrm{H}(Z)\right], 6.5\left[0.33 \mathrm{H}, \mathrm{s}, 1^{\prime}-\mathrm{H}(E)\right]\right\}$. Attempts to reduce the oxime with various standard reagents such as $\mathrm{LiAlH}_{4}$ or $\mathrm{BH}_{3}$ failed to provide the aminonucleoside. It was recognized that oxime $O$-acylation favoured reduction. Accordingly oxime (3a) was treated with acetic anhydride ( $\mathrm{Ac}_{2} \mathrm{O}$-pyridine-DMAP) then submitted to catalytic hydrogenation $\left(\mathrm{PtO}_{2}\right) .{ }^{16,17}$ Under these conditions, the nucleoside (5a), a polar compound giving a positive reaction with ninhydrin, was isolated in $20 \%$ yield. The presence of the amino group was indicated by the 60 MHz ${ }^{1} \mathrm{H}$ n.m.r. spectrum which had resonances at $\delta_{\mathrm{H}} 3.3$ (dd, $J 3.6$ and $5 \mathrm{~Hz})$ and $5.9(\mathrm{~d}, J 3.6 \mathrm{~Hz})$ assigned respectively to $1^{\prime}-\mathrm{H}$ and $2^{\prime}-$ H . These data indicated clearly that the addition of the hydride from the less hindered side led to the rare sugar L-talo- $2^{\prime}$-amino nucleoside (5a). Although the catalytic reduction furnished the amino derivative, the low yield prompted us to examine sodium borohydride reduction of oxime in the presence of nickel chloride or molybdenum(VI) oxide. ${ }^{18}$

Reaction of the hydroximino nucleoside (3a) with $\mathrm{NaBH}_{4}$ in methanol at $-30^{\circ} \mathrm{C}$ in the presence of nickel(II) chloride by means of the Ipaktschi procedure ${ }^{18}$ gave no conversion of the oxime group. In contrast treatment of the oxime- $O$-acetate (4a) afforded the $2^{\prime}$-amino nucleoside (5a) in $60 \%$ yield. Examination of the n.m.r. spectra at 60 MHz revealed the presence of two $2^{\prime}-$ amino nucleosides with the same, $J_{1,2}, 3.6 \mathrm{~Hz}$, coupling constants. Repeated purification showed only one product on t.l.c. and the compound homogeneity was ascertained by elemental analysis. Examination of the product at 300 MHz confirmed the presence of two l-talo- $2^{\prime}$-amino derivatives. However reduction of compound (4a) followed by treatment of the reaction mixture


(3) a; $R=$ Thyminyl $R^{1}=H$
b; $R=$ Theophyllinyl $R^{1}=H$
(4) a; $R=$ Thyminyl $R^{1}=$ MeCO
b; $R=$ Theophyllinyl $\mathbf{R}^{1}=$ MeCO

(5) a; $R=$ Thyminyl $R^{1}=H$
b; $R=$ Theophyllinyl $R^{1}=H$
(6) a; $R=$ Thyminyl $R^{1}=H$
(2) R = Thyminyl

(7) a; R $\begin{aligned} & =\text { Thyminyl } \\ \text { b; } R & =\text { Theophyllinyl }\end{aligned}$
(8) $\begin{aligned} & \mathbf{a} ; R=\text { Thyminyl } R^{1}=H \\ & \text { b; } R=\text { Theophyllinyl } R^{1}=H \\ &\text { ( }) ~ \text { a; } R \\ & \text { b; Thyminyl } R^{1}=C H C O \\ & \text { b; }=\text { Theophyllinyl } R^{1}=C H C O\end{aligned}$
$\begin{aligned} \text { (10) } \mathbf{a} ; & R=\text { Thyminyl } R^{1}=P h C H \text { OCO } \\ \text { b; } R & =\text { Theophyllinyl } R^{1}=\text { PhCH OCO } \\ \text { (11) } \mathbf{a} ; R & =\text { Thyminyl } R^{1}=H \\ \text { b; } R & =\text { Theophyllinyl } R^{1}=H \\ \text { (12) } \mathbf{a} ; R & =\text { Thyminyl } R^{1}=\text { MeCO } \\ \text { b; } R & =\text { Theophyllinyl } R^{1}=\text { MeCO }\end{aligned}$
with acetic anhydride yielded only the $2^{\prime}$-acetamido derivative (6a). Examination of the n.mr. spectrum revealed a single compound produced by the stereoselective addition of the hydride ion from the less hindered side. Finally acetylation of the crystalline compound (5a) with acetic anhydride in methanol gave compound (6a) quantitatively as a single product thus confirming the purity of the free amino nucleoside.*

The preparation of a $2^{\prime}$-amino purine derivative was carried out in a similar way. The theophylline nucleoside ${ }^{19}$ (1b) was treated with hydroxylamine hydrochloride. After 2 h compound (3b) was isolated as a single $Z$-isomer as evidenced by the downfield shift of the $1^{\prime}-\mathrm{H}$ signal at $\delta_{\mathrm{H}} 7.3$. Acetylation of compound ( $\mathbf{3 b}$ ) in the standard manner followed by borohydride reduction in the presence of $\mathrm{NiCl}_{2}$ furnished the amino nucleoside (5b) in $20 \%$ yield only. $\mathrm{MoO}_{3}$-Assisted reduction of (4b) at $0^{\circ} \mathrm{C}$ gave better results and afforded the l-talo

[^0]nucleoside ( $\mathbf{5 b}$ ) in $60 \%$ yield. The ${ }^{1} \mathrm{H}$ n.m.r. spectrum of this compound closely resembled that of compound (5a). At 300 MHz the presence of the axial amino group at $2^{\prime}-\mathrm{C}$ was indicated by the characteristic resonance at $\delta_{\mathrm{H}} 3.34(1 \mathrm{H}, \mathrm{dd}, J$ 2.6 and 5 Hz ) and by the $1^{\prime}-\mathrm{H}$ doublet at $\delta_{\mathrm{H}} 6.10(1 \mathrm{H}, \mathrm{d}, J 2.6 \mathrm{~Hz})$.

This success prompted further exploration and our attention was focussed on $4^{\prime}$-keto nucleosides. The keto nucleosides $(7 a)^{20}$ and (7b) ${ }^{21}$ were treated with hydroxylamine hydrochloride affording oximes (8a) and (8b) as the $E$-isomers. The cis relationship between the oxime and the $5^{\prime}-\mathrm{C}$ carbon was clearly discernible by the upfield shift of the $5^{\prime}-\mathrm{H}$ resonance at $\delta_{\mathrm{H}} 5.3$ for (8a) and 5.04 for (8b).
Reduction of the acetylated oximes (9a) and (9b) in the presence of nickel chloride gave the $4^{\prime}$-amino nucleosides which were isolated as benzyl carbamates (10a) and (10b). Catalytic reduction of compounds (10a) and (10b) with $\mathrm{Pd}-\mathrm{C}$ afforded the free amino nucleosides (11a) and (11b) but treatment with $\mathrm{MoO}_{3}-\mathrm{NaBH}_{4}$ gave the cleanest reaction allowing direct isolation of (11a) and (11b). The structures of the products were deduced from the ${ }^{1} \mathrm{H}$ n.m.r. spectra of the $4^{\prime}$-acetamido derivatives ( $\mathbf{1 2 a}$ ) and ( $\mathbf{1 2 b}$ ). The presence of the $4^{\prime}$-amido group was indicated by the $4^{\prime}-\mathrm{H}$ signals [(12a) $\delta_{\mathrm{H}} 4.67(1 \mathrm{H}$, ddd, $J 4.4$, 6.7 , and 8.9 Hz$)$, (12b) $\delta_{\mathrm{H}} 4.75(1 \mathrm{H}$, ddd, $J 4.7,6.4$, and 8.9 Hz$)$ ]. Examination of the $J_{4^{\prime}, 5}$, coupling constants [(12a) 6.7 Hz , (12b) $6.4 \mathrm{~Hz}]$ suggested an axial equatorial relationship indicating addition of the hydride from the less hindered side. On the basis of molecular models it was expected that NOESY data would
ascertain the stereochemistry at $4^{\prime}-C$. Thus 2D proton-proton correlated spectra using CONOESY pulse sequence were recorded. For both the $4^{\prime}$-acetamido nucleosides (12a) and (12b) the $6^{\prime}-\mathrm{H}$ protons show a n.O.e. interaction with $1^{\prime}-\mathrm{H}$ and NH. Furthermore no correlation could be established between NH, $3^{\prime}-\mathrm{H}$, and $5^{\prime}-\mathrm{H}$. The $J_{\mathrm{NH}, 4^{\prime}-\mathrm{H}}$ value of 8.9 Hz suggesting a trans relationship, ${ }^{22}$ these above observations demonstrated a syn relationship between the $6^{\prime}-\mathrm{Me}, 1^{\prime}-\mathrm{H}$, and the $4^{\prime}$-acetamido group consistent with an l-talo configuration for compounds (12a) and (12b).

Reduction of $x, \beta$-unsaturated oximes ${ }^{23}$ was also investigated in an attempt to prepare $2^{\prime}, 3^{\prime}$-unsaturated $4^{\prime}$-amino nucleosides. Treatment of the keto nucleoside (13) with hydroxylamine



hydrochloride resulted in the exclusive formation of the oxime (14a). The conjugated nature of (14a) was confirmed by the characteristic ${ }^{1} \mathrm{H}$ n.m.r. spectrum consisting of resonances at $\delta_{\mathbf{H}}$ $4.8\left(\mathrm{q}, J 6.5 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 6.5\left(J 1.8\right.$ and $\left.10 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right)$, and $7.5(J 1.8$ and $\left.10 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right)$. Acetylation of the oxime (14a) yielded the oximinyl acetate (14b).

Reduction of compound (14b) with $\mathrm{NaBH}_{4}-\mathrm{NiCl}_{2}$ afforded a 1:1 mixture of two labile amines which could not be isolated. Treatment of the reaction mixture with benzyl chloroformate afforded the benzyl carbamates (15a) and (15b). As expected the presence of four methylene protons at $\delta_{\mathrm{H}} 1.81,2.15$, and 2.25 p.p.m. for the less polar isomer (15a) and at $\delta_{\mathrm{H}} 1.95$ and $\delta_{\mathrm{H}} 2.48$ for ( $\mathbf{1 5 b}$ ) indicated the reduction of the double bond. Furthermore examination of the $5^{\prime}-\mathrm{H}$ signal revealed a $J_{4^{\prime}}, 5^{\prime} 9.89 \mathrm{~Hz}$ coupling for the 7-(4'-benzyloxycarbonylamino- $\mathbf{2}^{\prime}, 3^{\prime}, 4^{\prime}, 6^{\prime}$-tetra-deoxy- $\beta$-L-erythro-hexopyranosyl)theophylline (15a) and a 1.6 Hz coupling constant for the 7-(4'-benzyloxycarbonylamino-
$2^{\prime}, 3^{\prime}, 4^{\prime}, 6^{\prime}$-tetradeoxy- $\beta$-L-threo-hexopyranosyl)theophylline (15b).

Molybdenum oxide assisted reduction of the oxime (14b) also afforded two products. Crystallization of the less polar isomer afforded the amino nucleoside (16a). The unsaturated nature of (16a) was deduced from the ${ }^{1} \mathrm{H}$ n.m.r. spectrum with the two olefinic multiplets occurring at $\delta_{\mathrm{H}} 5.92$ and 6.15 p.p.m. Examination of the $5^{\prime}-\mathrm{H}$ resonance at $\delta_{\mathrm{H}} 3.71$ (dq, $J 6.1$ and 8.36 Hz ) indicated an axial addition of the hydride ion leading to the L-erythro- $4^{\prime}$-amino nucleoside.

The 300 MHz high resolution n.m.r. spectrum of compound (17) was consistent with the $4^{\prime}$-amino tetrahydropyranosyl structure. The detailed assignments were supported by two dimensional chemical shift correlation. The high resolution COSY spectrum of compound (17) (Figure 1) clearly indicated that the $1^{\prime}-\mathrm{H}$ protons at $\delta_{\mathrm{H}} 5.98$ p.p.m. is coupled with the signal at $\delta_{\mathrm{H}} 1.84$ and 2.30 assigned respectively to $2^{\prime}-\mathrm{H}_{\mathrm{a}}$ and $2^{\prime}-\mathrm{H}_{\mathrm{e}}$ and that these two protons are coupled with each other. These two signals are correlated with the $3^{\prime}-\mathrm{H}$ resonances at $\delta_{\mathrm{H}} 1.58\left(3^{\prime}-\mathrm{H}_{\mathrm{a}}\right)$ and $\delta_{\mathrm{H}} 2.12\left(3^{\prime}-\mathrm{H}_{\mathrm{e}}\right)$ which in turn are coupled with each other and with the $4^{\prime}-\mathrm{H}$ signal at $\delta_{\mathrm{H}} 2.58 \mathrm{~Hz}$. Finally the overlapped resonance at $\delta_{\mathrm{H}} 3.46$ correlated to $4^{\prime}-\mathrm{H}$ and $6^{\prime}-\mathrm{H}$ was assigned to $5^{\prime}-\mathrm{H}$. The $4^{\prime}-\mathrm{H}$ proton appeared as a multiplet ( 1 H , ddd, $J 4.05$, 9.25 , and 11.17 Hz ) with a large $J_{4^{\prime}, 5^{\prime}} 9.25 \mathrm{~Hz}$ coupling constant consistent with the l-erythro configuration for the labile amine (16a). Direct treatment of the reaction mixture with benzyl chloroformate furnished the 7-(4'-benzylox ycarbonylaminohex-$2^{\prime}$-enopyranosyl)theophylline (16b) and a saturated carbamate which proved to be identical with the 7-(4'-benzyloxycarbonyl-amino- $2^{\prime}, 3^{\prime}, 4^{\prime}, 6^{\prime}$-tetradeoxy- $\beta$-L-erythro-hexopyranosyl)theophylline (15a) prepared by nickel chloride assisted reduction.

We have devised a new method for the preparation of amino hexopyranose nucleosides from keto nucleosides and found that these reactions are highly stereoselective and lead to amino nucleosides by hydride addition from the less hindered side. The formation of allylic amino nucleosides is also stereoselective and affords the equatorial amino group suggesting that this methodology could be applied to the synthesis of naturally occurring nucleosides.

## Experimental

${ }^{1}$ H N.m.r. spectra were recorded in the indicated solvent on a Varian T60 ( 60 MHz ) or a Bruker MSL $300(300.13 \mathrm{MHz})$ instrument. Chemical shifts are reported in p.p.m. downfield from tetramethylsilane with coupling constants in Hertz. Two dimensional $J$ correlated CONOESY n.m.r. experiments were performed at 300.13 MHz using the program supplied in the manufacturer's software. A mixing time of 400 ms was used to collect 256 transients for each of the 64 spectra. The high resolution COSY experiment was carried out according to reference 24. 512 Experiments of 16 transients were recorded. The relaxation delay was 1 s and the digital resolution along both axes was $1.53 \mathrm{~Hz} /$ point. I.r. spectra were obtained on a Perkin-Elmer 137 spectrophotometer and u.v. spectra on a Varian 635 spectrophotometer. Microanalyses were performed by the Laboratoire central de Microanalyse du CNRS, Vernaison, France. T.l.c. was carried out on E. Merck AG Darmstad F254 silicagel pre-coated plastic sheet. The term flash chromatography is used for medium pressure liquid chromatography (Silicagel $60,0.04-0.063 \mathrm{~mm}$ ).
(E,Z)-1-(2'-Acetoximino-2', $6^{\prime}$-dideoxy- $3^{\prime}, 4^{\prime}$-O-isopropylidene-$\beta$-L-talo-hexopyranosyl)thymine (4a).—Pyridine ( $0.492 \mathrm{~g}, 6.24$ $\mathrm{mmol})$, acetic acid anhydride ( $0.636 \mathrm{~g}, 6.24 \mathrm{mmol}$ ) and several crystals of 4-dimethylaminopyridine were added to a solution of ( $E, Z$ )-1-( $2^{\prime}, 6^{\prime}$-dideoxy- $3^{\prime}, 4^{\prime}$ - $O$-isopropylidene-2'-hydroximino-


Figure 1. Contour plot of a high resolution COSY spectrum for (17). The $1 \mathrm{D} 300 \mathrm{MHz}{ }^{1} \mathrm{H}$ n.m.r. spectrum is shown above.
$\beta$-L-lyxo-hexopyranosyl)thymine (3a)* $(1.688 \mathrm{~g}, 5.2 \mathrm{mmol})$ in dichloromethane ( 52 ml ). After being stirred for 2 h the resulting mixture was concentrated under reduced pressure. Crystallisation yielded compound (4a) as a $2: 1$ mixture of $(E)$ and $(Z)$ isomers ( $1.465 \mathrm{~g}, 80 \%$ ).

[^1]Mixture: m.p. $177-180^{\circ} \mathrm{C}$ (from EtOH); $[\alpha]_{\mathrm{D}}^{20}-35^{\circ}(c 0.1$ in MeOH ); $\lambda_{\text {max }}$. MeOH ) 262 nm ( $\varepsilon$ in $\mathrm{dm}^{3} \mathrm{~mol}^{-1} \mathrm{~cm}^{-1} 6973$ ); $\delta_{\mathrm{H}}\left[60 \mathrm{MHz},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right] 1.30\left(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 1.45$ and $1.60\left(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 1.80(3 \mathrm{H}, \mathrm{s}, 6-\mathrm{Me}), 2.10(3 \mathrm{H}, \mathrm{s}$, $\mathrm{MeCO}), 4\left(1 \mathrm{H}, \mathrm{dq}, J 1.5\right.$ and $\left.6 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 4.20(1 \mathrm{H}, \mathrm{dd}, J 1.5$ and $\left.7 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 4.80$ and $5.40(0.33 \mathrm{H}$ and 0.66 H , two d, $J 7 \mathrm{~Hz}$, $\left.3^{\prime}-\mathrm{H}\right), 6.36$ and $6.46\left(0.66 \mathrm{H}\right.$, and $\left.0.33 \mathrm{H}, 1^{\prime}-\mathrm{H}\right)$, and 7.50 and 7.80 ( 0.66 H and $0.33 \mathrm{H}, 2 \times \mathrm{s}, 6-\mathrm{H}$ ).
'Flash' chromatography on silica gel with ethyl acetatepentane ( $6: 4$ ) then crystallisation afforded the pure derivatives.
(E): m.p. $212{ }^{\circ} \mathrm{C}$ (from EtOH); $[\alpha]_{\mathrm{D}}^{20}-60^{\circ}(c 0.1$ in MeOH)
(Found $\mathrm{C}, 52.2 ; \mathrm{H}, 5.65 ; \mathrm{N}, 11.45 . \mathrm{C}_{16} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{7}$ requires C , $52.32 ; \mathrm{H}, 5.72 ; \mathrm{N}, 11.34 \%$ ); $\lambda_{\text {max. }}(\mathrm{MeOH}) 262 \mathrm{~nm}(\varepsilon 7523)$.
( $Z$ ): m.p. $192{ }^{\circ} \mathrm{C}$ (from EtOH); $[\alpha]_{\mathrm{D}}^{20}-20^{\circ}(c 0.1$ in MeOH$)$ (Found: C, 52.2; H, 5.7; N, 11.3); $\lambda_{\text {max }}(\mathrm{MeOH}) 262 \mathrm{~nm}(\varepsilon 8881)$.

1-( $2^{\prime}$-Amino- $2^{\prime}, 6^{\prime}$-dideoxy- $3^{\prime}, 4^{\prime}$ - O -isopropylidene- $\beta$-L-talohexopyranosyl)thymine (5a).-Nickel chloride ( $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}$; $1.6 \mathrm{~g}, 6.75 \mathrm{mmol}$ ) was added to a solution of the ( $E, Z$ ) nucleoside (4a) ( $1.04 \mathrm{~g}, 2.8 \mathrm{mmol}$ ) in methanol ( 30 ml ) at $-30^{\circ} \mathrm{C}$. To the resulting suspension sodium borohydride (1.06 $\mathrm{g}, 28 \mathrm{mmol}$ ) was added slowly. The reaction mixture was stirred for 1 h . It was then poured into brine ( 25 ml ) and ethyl acetate $(25 \mathrm{ml})$ and the aqueous phase was extracted with ethyl acetate $(3 \times 25 \mathrm{ml})$. The combined organic extracts were dried ( $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ) then evaporated under reduced pressure. Column chromatography on silica gel with ethyl acetate-methanol (8:2) gave the $2^{\prime}$-amino-talo-nucleoside ( 5 a ) ( $0.522 \mathrm{~g}, 60 \%$ ), m.p. $130^{\circ} \mathrm{C}$ (from acetone); $[\alpha]_{\mathrm{D}}^{20}-80^{\circ}(c 0.1$ in MeOH$)$ (Found: C , $52.65 ; \mathrm{H}, 6.5 ; \mathrm{N}, 13.3 . \mathrm{C}_{14} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{5} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 52.5 ; \mathrm{H}$, $6.87 ; \mathrm{N}, 13.2 \%$ ); $\lambda_{\text {max }}(\mathrm{MeOH}) 266 \mathrm{~nm}(\varepsilon 11133) ; \delta_{\mathrm{H}}(300.13$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.39\left(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 1.52\left(1.3 \mathrm{H}, \mathrm{d}, J 6.8 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right)$, $1.6\left(1.7 \mathrm{H}, \mathrm{d}, J 6.8 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 1.63$ and $1.67(1.3 \mathrm{H}$ and 1.7 H , $2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}$ ), 1.94 and 1.98 ( 1.3 H and $1.7 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me} 5-\mathrm{C}$ ), $3.48\left(0.43 \mathrm{H}\right.$, dd, $\left.J 3.6 \mathrm{and} 4.3 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right), 4.06(0.43 \mathrm{H}$, dd, $J 2.6$ and $\left.6.8 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 4.11-4.2\left(\mathrm{~m}, 1.57 \mathrm{H}, 4^{\prime}-\mathrm{H}\right.$ and $\left.5^{\prime}-\mathrm{H}\right), 4.24$ ( $0.57 \mathrm{H}, J 3.6$ and $5.7 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}$ ), $4.49(0.43 \mathrm{H}$, dd, $J 4.3$ and 5.7 $\left.\mathrm{Hz}, 3^{\prime}-\mathrm{H}\right), 4.64\left(0.57 \mathrm{H}\right.$, dd, $J 5.7$ and $\left.6.1 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 6.15(0.43 \mathrm{H}$, $\left.\mathrm{s}, J 3.6 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right), 6.31\left(0.57 \mathrm{H}, \mathrm{s}, J 3.6 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right)$, and 8.1 and 8.16 ( 0.43 H and $0.57 \mathrm{H}, 2 \times \mathrm{s}, 6-\mathrm{H}$ ).

## 1-( $2^{\prime}$-Acetamido- $2^{\prime}, 6^{\prime}$-dideoxy- $3^{\prime}, 4^{\prime}$-O-isopropylidene- $\beta$ - L -

 talo-hexopyranosyl)thymine (6a).-(a) From the oxime acetate (4a). Nickel chloride $\left(\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O} ; 0.8 \mathrm{~g}, 3.36 \mathrm{mmol}\right)$ was added to a solution of the nucleoside ( 4 a ) $(0.52 \mathrm{~g}, 1.42 \mathrm{mmol})$ in methanol ( 15 ml ) at $-30^{\circ} \mathrm{C}$. To the resulting suspension sodium borohydride ( $0.53 \mathrm{~g}, 14 \mathrm{mmol}$ ) was added slowly. The reaction mixture was stirred for 1 h then acetic anhydride ( 0.33 $\mathrm{g}, 3.47 \mathrm{mmol}$ ) was added. The reaction mixture was then poured into brine ( 15 ml ) and ethyl acetate ( 15 ml ) and the aqueous phase was extracted with ethyl acetate $(3 \times 15 \mathrm{ml})$. The combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated under reduced pressure to afford the $2^{\prime}$-acetamido-talonucleoside ( 6 a ) ( $0.247 \mathrm{~g}, 50 \%$ ).(b) From the amino nucleoside (5a). Acetic anhydride was added $(0.121 \mathrm{~g}, 1.2 \mathrm{mmol})$ to a solution of the nucleoside ( 5 a ) $(0.311 \mathrm{~g}, 1 \mathrm{mmol})$ in methanol $(10 \mathrm{ml})$ at $25^{\circ} \mathrm{C}$. After 30 min the solvent was evaporated under reduced pressure to give the $2^{\prime}-$ acetamido-talo-nucleoside ( 6 a ) $\left(0.47 \mathrm{~g}, 95 \%\right.$ ), m.p. $145-148{ }^{\circ} \mathrm{C}$ (from diethyl ether-ethanol), $[\alpha]_{\mathrm{D}}^{20}-85^{\circ} \mathrm{C}(c 0.1 \mathrm{inMeOH})$ (Found: C, 51.25; $\mathrm{H}, 6.55 ; \mathrm{N}, 10.9 . \mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{6} \cdot \mathrm{H}_{2} \mathrm{O}$ requires C, $51.75 ; \mathrm{H}, 6.73 ; \mathrm{N}, 11.32 \%$ ); $\lambda_{\text {max. }}$. (MeOH) $266 \mathrm{~nm}(\varepsilon 11260)$; $\delta_{\mathrm{H}}\left(60 \mathrm{MHz}, \mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}\right) 1.2-1.4\left(9 \mathrm{H}, \mathrm{m}, J 6.5 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right.$ and $\mathrm{Me}_{2} \mathrm{C}$ ), 1.8 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO}$ ), 2.1 ( $3 \mathrm{H}, \mathrm{s}, 5-\mathrm{Me}$ ), $3.8(1 \mathrm{H}, \mathrm{dq}, J 2$ and $\left.6.5 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 4.1\left(1 \mathrm{H}\right.$, dd, $J^{2}$ and $\left.7 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 4.6(1 \mathrm{H}$, dd, $J$ 5 and $\left.7 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right)$, $5.2\left(1 \mathrm{H}\right.$, ddd, $J 5,5$, and $\left.8.7 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right), 6.4(1$ $\left.\mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right), 7.9(1 \mathrm{H}, \mathrm{s}, 6-\mathrm{H})$, and $8.4(1 \mathrm{H}, \mathrm{d}, J 8.7 \mathrm{~Hz}$, NH).
(Z)-7-(2',6'-Dideoxy-2'-hydroximino-3', $4^{\prime}$-O-isopropylidene-$\beta$-L-lyxo-hexopyranosyl)theophylline (3b).-Hydroxylamine hydrochloride ( $1.4 \mathrm{~g}, 20 \mathrm{mmol}$ ) was added to a solution of 7 ( $6^{\prime}$-deoxy- $3^{\prime}, 4^{\prime}$ - $O$-isopropylidene - $\beta$-L-lyxo-hexopyranosulosyl) theophylline ( $\mathbf{1 b}$ ) $(3.64 \mathrm{~g}, 10 \mathrm{mmol})$ in a $1: 1$ ethanol-pyridine mixture. The mixture was stirred at room temperature for 4 h then water was added and the aqueous mixture was extracted with dichloromethane $(3 \times 15 \mathrm{ml})$. The dichloromethane extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ to leave, after the solvent was evaporated off, a crude product from which was isolated the oxime (3b) ( $3.03 \mathrm{~g}, 80 \%$ ), m.p. $150^{\circ} \mathrm{C}$ (from MeOH ); $[\alpha]_{\mathrm{D}}^{20}$
$-55^{\circ}(c 0.1$ in MeOH$)$ (Found: C, $49.35 ; \mathrm{H}, 5.5$; $\mathrm{N}, 18.05$. $\mathrm{C}_{16} \mathrm{H}_{21} \mathrm{~N}_{5} \mathrm{O}_{6} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 49.48 ; \mathrm{H}, 5.67 ; \mathrm{N}, 18.08 \%$ ); $\lambda_{\text {max. }}(\mathrm{MeOH}) 276 \mathrm{~nm}(\varepsilon 9209) ; \delta_{\mathrm{H}}\left[60 \mathrm{MHz},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right] 1.3$ (3 $\left.\mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 1.5$ and $1.6\left(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 3.4$ and $3.6\left(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}\right.$, NMe), $3.8\left(1 \mathrm{H}, \mathrm{dq}, J 7\right.$ and $\left.2 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 4.5$ $\left(1 \mathrm{H}, \mathrm{dd}, J 2\right.$ and $\left.8 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 5.1\left(1 \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 7.3(1 \mathrm{H}, \mathrm{s}$, $\left.1^{\prime}-\mathrm{H}\right)$, and $8.1(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.

7-(2'-Acetoximino- $2^{\prime}, 6^{\prime}$-dideoxy- $3^{\prime}, 4^{\prime}$ - O -isopropylidene- $\beta-\mathrm{L}-$ hexopyranosyl)theophylline (4b). -The hydroximinyl nucleoside (3b) $(1.875 \mathrm{~g}, 5 \mathrm{mmol})$ in pyridine $(10 \mathrm{ml})$ was treated with acetic anhydride ( $0.61 \mathrm{~g}, 6 \mathrm{mmol}$ ) in the presence of a catalytic amount of 4-dimethylaminopyridine. The resulting solution was worked up as for (4a) to yield the title compound (4b); m.p. $205^{\circ} \mathrm{C}$ (from EtOH); [ $\alpha]_{\mathrm{D}}^{20}-125^{\circ}(c 0.1$ in MeOH) (Found: C, 51.45 ; H, 5.5 ; $\mathrm{N}, 16.45 . \mathrm{C}_{18} \mathrm{H}_{23} \mathrm{~N}_{5} \mathrm{O}_{7}$ requires C, $51.30 ; \mathrm{H}, 5.46 ; \mathrm{N}, 16.62 \%$ ); $\lambda_{\text {max. }}$. MeOH ) $276 \mathrm{~nm}(\varepsilon 7990) ; \delta_{\mathrm{H}}\left(60 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.3(3 \mathrm{H}, \mathrm{d}$, $\left.J 7 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 1.5$ and $1.6\left(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 2(3 \mathrm{H}, \mathrm{s}$, MeCO ), 3.4 and $3.6\left(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}\right.$, NMe), $3.7\left(1 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}\right)$, $4.5\left(1 \mathrm{H}, \mathrm{dd}, J 2\right.$ and $\left.8 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 5.1\left(1 \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 7.3$ ( 1 $\left.\mathrm{H}, \mathrm{s}, 1^{\prime}-\mathrm{H}\right)$, and $8.1(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.

7-( $2^{\prime}$-Amino- $2^{\prime}, 6^{\prime}$-dideoxy- $3^{\prime}, 4^{\prime}$ - O -isopropylidene- $\beta$-L-talohexopyranosyl) theophylline (5b).-Molybdenum oxide ( 0.36 g , $2.5 \mathrm{mmol})$ was added to a solution of the nucleoside (3b) ( 0.421 $\mathrm{g}, 1 \mathrm{mmol})$ in $\mathrm{MeOH}(10 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$. Sodium borohydride $(0.76$ $\mathrm{g}, 20 \mathrm{mmol}$ ) was added portionwise for 1 h then the reaction mixture was poured into brine $(10 \mathrm{ml})$ and ethyl acetate $(10 \mathrm{ml})$ and the aqueous solution was extracted with ethyl acetate $(3 \times 10 \mathrm{ml})$. The combined organic extracts were dried ( $\mathrm{Na}_{2}-$ $\mathrm{SO}_{4}$ ) then evaporated under reduced pressure. Column chromatography on silica gel (ethyl acetate- $\mathrm{MeOH}, 9: 1$ ) gave the $2^{\prime}-$ amino-talo-nucleoside (5b) ( $0.219 \mathrm{~g}, 60 \%$ ), m.p. $195-197^{\circ} \mathrm{C}$ (from diethyl ether-pentane); $[\alpha]_{\mathrm{D}}^{20}-95^{\circ}(c 0.1$ in MeOH$)$ (Found: C, 52.6; H, 6.4; N, 19.9. $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{5} \mathrm{O}_{5}$ requires C, 52.60 ; $\mathrm{H}, 6.30 ; \mathrm{N}, 19.17 \%$ ); $\lambda_{\text {max. }}(\mathrm{MeOH}) 275 \mathrm{~nm}(\varepsilon 9563)$; $\delta_{\mathrm{H}}(300.13$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.44\left(3 \mathrm{H}, \mathrm{d}, J 6.3 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 1.41$ and $1.60(2 \times 3$ $\mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}$ ), $3.34\left(1 \mathrm{H}, \mathrm{dd}, J 2.6\right.$ and $\left.5 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right), 3.4$ and 3.6 $(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{NMe}), 4.10-4.20\left(2 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}\right.$ and $\left.5^{\prime}-\mathrm{H}\right), 4.55$ $\left(1 \mathrm{H}, \mathrm{dd}, J 5\right.$ and $\left.6.1 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 6.10\left(1 \mathrm{H}, \mathrm{d}, J 2.6 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right)$, and 8.43 ( $1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H}$ ).
(E)-1-(4', $6^{\prime}$-Dideoxy-4'-hydroximino-2', $3^{\prime}$-O-isopropylidene-$\alpha$-L-lyxo-hexopyranosyl)thymine (8a).-Hydroxylamine hydrochloride $(2.105 \mathrm{~g}, 30.3 \mathrm{mmol})$ was added to a solution of 1-( $6^{\prime}-$ deoxy- $2^{\prime}, 3^{\prime}-O$-isopropylidene- $\alpha$-L-lyxo-hexopyranos-4'-ulosyl)thymine ( $7 \mathbf{a}$ ) $(4.7 \mathrm{~g}, 15.15 \mathrm{mmol})$ in ethanol ( 15 ml ) and pyridine $(15 \mathrm{ml})$. After 2 h the solution was poured into brine $(30 \mathrm{ml})$. The aqueous phase was extracted with ethyl acetate $(3 \times 30 \mathrm{ml})$ and the combined extracts were evaporated to yield (8a) ( 4.43 g , $90 \%$ ), m.p. $303{ }^{\circ} \mathrm{C}$ (from EtOH ); $[\alpha]_{\mathrm{D}}^{20}-40^{\circ}(c 0.1$ in MeOH ) (Found: C, $51.8 ; \mathrm{H}, 5.8 ; \mathrm{N}, 13.1 . \mathrm{C}_{14} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{6}$ requires $\mathrm{C}, 51.68$; $\mathrm{H}, 5.68 ; \mathrm{N}, 12.91 \%$ ); $\lambda_{\text {max. }}(\mathrm{MeOH}) 264 \mathrm{~nm}(\varepsilon 9845) ; \delta_{\mathrm{H}}(60 \mathrm{MHz}$, $\left.\mathrm{CD}_{3} \mathrm{CO}_{2} \mathrm{D}\right) 1.55\left(3 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 1.45$ and $1.60(2 \times 3 \mathrm{H}$, $\left.2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 1.95(3 \mathrm{H}, \mathrm{s}, 5-\mathrm{Me}), 4.6\left(1 \mathrm{H}, \mathrm{dd}, J 6\right.$ and $7 \mathrm{~Hz}, 2^{\prime}-$ H), $5\left(1 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 5.3\left(\mathrm{q}, 1 \mathrm{H}, J 7 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 5.95(1 \mathrm{H}, J$ $\left.7 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right)$, and $7.5(1 \mathrm{H}, \mathrm{s}, 6-\mathrm{H})$.
(E)-1-(4'-Acetoximino-4', $6^{\prime}$-dideoxy- $2^{\prime}, 3^{\prime}$-isopropylidene- $\alpha-\mathrm{L}-$ lyxo-hexopyranosyl)thymine (9a).-A 25 ml round bottom flask was charged successively with $4^{\prime}$-hydroximinyl nucleoside (8a) $(8.44 \mathrm{~g}, 26 \mathrm{mmol})$, pyridine ( $2.46 \mathrm{~g}, 31.2 \mathrm{mmol}$ ), acetic anhydride ( $3.18 \mathrm{~g}, 31.2 \mathrm{mmol}$ ), dichloromethane ( 26 ml ), and a catalytic amount of 4-dimethylaminopyridine. After being stirred for 2 h the solution was worked up as for (4a) to yield the title compound (9a) ( $8.58 \mathrm{~g}, 90 \%$ ), m.p. $132-135^{\circ} \mathrm{C}$ (from EtOH ); $[\alpha]_{\mathrm{D}}^{20}-25^{\circ}(c 0.1$ in MeOH ) (Found: C, 52.1; H, 5.9; N, 11.4. $\mathrm{C}_{16} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{7}$ requires $\mathrm{C}, 52.31 ; \mathrm{H}, 5.72 ; \mathrm{N}, 11.39 \%$;
$\lambda_{\text {max. }}(\mathrm{MeOH}) 264(\varepsilon 9615) ; \delta_{\mathbf{H}}\left(300.13 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.39$ and $1.60\left(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me}_{3} \mathrm{C}\right), 1.62\left(3 \mathrm{H}, \mathrm{d}, J 6.8 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 1.92(3$ $\mathrm{H}, \mathrm{s}, 5-\mathrm{Me}), 2.22(3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO}), 4.50(1 \mathrm{H}, \mathrm{dd}, J 5.7$ and 6.8 Hz , $\left.2^{\prime}-\mathrm{H}\right), 5.03\left(1 \mathrm{H}, \mathrm{dd}, J 5.7 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 5.10\left(\mathrm{q}, 1 \mathrm{H}, J 6.8 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right)$, $5.86\left(1 \mathrm{H}, \mathrm{dd}, J 6.8 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right)$, and $7.07(1 \mathrm{H}, \mathrm{s}, 6-\mathrm{H})$.

1-(4'-Benzyloxycarbonylamino-4', $6^{\prime}$-dideoxy $\mathbf{- ~}^{\prime}, 3^{\prime}$-O-iso-propylidene-x-L-talo-hexopyranosyl)thymine (10a).-Nickel chloride ( $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O} ; 0.57 \mathrm{~g}, 2 \mathrm{mmol}$ ) was added to a solution of the nucleoside ( 9 a ) $(0.367 \mathrm{~g}, 1 \mathrm{mmol})$ in methanol ( 10 ml ) at $-30^{\circ} \mathrm{C}$. To the resulting suspension sodium borohydride $(0.38$ $\mathrm{g}, 10 \mathrm{mmol}$ ) was added slowly. After being stirred for 0.5 h , the reaction mixture was cooled to $-60^{\circ} \mathrm{C}$ and benzyl chloroformate ( $1.195 \mathrm{~g}, 7 \mathrm{mmol}$ ) was added. The reaction mixture was warmed to $0^{\circ} \mathrm{C}$ and stirred for 1 h , then poured into brine ( 10 $\mathrm{ml})$ and ethyl acetate $(10 \mathrm{ml})$. The aqueous phase was extracted with ethyl acetate ( $3 \times 10 \mathrm{ml}$ ) and the combined organic extracts dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated under reduced pressure. The residual oil washed with pentane to remove the benzyl derivative gave compound (10a) $(0.266 \mathrm{~g}, 60 \%)$, m.p. $238^{\circ} \mathrm{C}$ (from EtOH ); $[x]_{\mathrm{D}}^{20}+5^{\circ}(c 0.1$ in MeOH ) (Found: C , 59.45; $\mathrm{H}, 6.1 ; \mathrm{N}, 9.5 . \mathrm{C}_{22} \mathrm{H}_{26} \mathrm{~N}_{3} \mathrm{O}_{7}$ requires $\mathrm{C}, 59.45 ; \mathrm{H}, 5.85 ; \mathrm{N}$, $9.45 \%) ; \lambda_{\text {max }}(\mathrm{MeOH}) 264 \mathrm{~nm}(\varepsilon 7636) ; \delta_{\mathrm{H}}[300.13 \mathrm{MHz}$, $\left.\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right] 1.41\left(3 \mathrm{H}, \mathrm{d}, J 6.4 \mathrm{~Hz}, \mathrm{~b}^{\prime}-\mathrm{H}\right), 1.33$ and $1.54(2 \times 3 \mathrm{H}$, $\left.2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 1.92(3 \mathrm{H}, \mathrm{s}, 5-\mathrm{Me}), 4.32-4.41\left(2 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}\right.$ and $\left.5^{\prime}-\mathrm{H}\right), 4.5\left(1 \mathrm{H}\right.$, dd, $J 4.7$ and $\left.7 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right), 4.63(1 \mathrm{H}, \mathrm{dd}, J 4.6$ and $\left.4.7 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 5.17-5.21\left(2 \times 1 \mathrm{H}, 2 \times \mathrm{d}, J 6 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}\right)$, $5.92\left(1 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right)$, and $7.45(1 \mathrm{H}, \mathrm{s}, 6-\mathrm{H})$.

1-(4'-Amino-4', $6^{\prime}$-dideoxy- $2^{\prime}, 3^{\prime}$-O-isopropylidene- $\alpha$ - L -lyxohexopyranosyl)thymine (11a).-(a) From benzyl carbamate (10a). A 125 ml high pressure reactor was charged with compound ( 10 aa ) ( $0.444 \mathrm{~g}, 1 \mathrm{mmol}$ ), methanol ( 20 ml ) and $\mathrm{Pd}-\mathrm{C}$ catalyst ( 0.044 g ). The reactor was pressurized with hydrogen at 3 atm . After 1 h the suspension was filtered off. The solvent was evaporated off and the aminonucleoside (11a) ( $0.248 \mathrm{~g}, 80 \%$ ) was crystallized from ethanol.
(b) From oxime (9a). Molybdenum oxide ( $0.36 \mathrm{~g}, 2.5 \mathrm{mmol}$ ) was added to a solution of the nucleoside (9a) $(0.367 \mathrm{~g}, 1 \mathrm{mmol})$ in $\mathrm{MeOH}(10 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$. Sodium borohydride $(0.76 \mathrm{~g}, 20$ mmol ) was added portionwise for 1 h then the reaction mixture was poured into brine $(10 \mathrm{ml})$ and ethyl acetate $(10 \mathrm{ml})$, and the aqueous solution was extracted with ethyl acetate $(10 \mathrm{ml})$. The combined organic extracts were dried ( $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ), then evaporated under reduced pressure to give after crystallization the $4^{\prime}$ amino nucleoside (11a) ( $0.153 \mathrm{~g}, 50 \%$ ), m.p. $182^{\circ} \mathrm{C}$ (from EtOH); $[x]_{\mathrm{D}}^{20}-30^{\circ}(c 0.1$ in MeOH ) (Found: C, $53.85 ; \mathbf{H}, 7.35 ; \mathrm{N}, 11.6$. $\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{~N}_{3} \mathrm{O}_{5} \cdot \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}$ requires $\mathrm{C}, 53.78 ; \mathrm{H}, 7.56 ; \mathrm{N}, 11.76 \%$ ); $\lambda_{\text {max. }}(\mathrm{MeOH}) 264 \mathrm{~nm}(\varepsilon 11594) ; \delta_{\mathrm{H}}\left(300.13 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.4$ (3 $\left.\mathrm{H}, \mathrm{d}, J 6.8 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 1.36$ and $1.60\left(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 1.91$ ( $3 \mathrm{H}, \mathrm{s}, 5-\mathrm{Me}$ ), $3.35\left(1 \mathrm{H}, \mathrm{dd}, J 4.6\right.$ and $4.8 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}$ ), $4.17(1 \mathrm{H}$, $\mathrm{dq}, J 4.6$ and $\left.6.8 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 4.22\left(1 \mathrm{H}, \mathrm{dd}, J 5.8\right.$ and $\left.8 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right)$, $4.48\left(1 \mathrm{H}, \mathrm{dd}, J 4.8\right.$ and $\left.5.8 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 5.94\left(1 \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right)$, and $7.5(1 \mathrm{H}, \mathrm{s}, 6-\mathrm{H})$.

1-(4'-Acetamido-4', $6^{\prime}$-dideoxy- $2^{\prime}, 3^{\prime}$-isopropylidene- $\alpha-\mathrm{L}$-talohexopyranosyl)thymine (12a). $-\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(0.57 \mathrm{~g}, 2 \mathrm{mmol})$ and sodium borohydride $(0.38 \mathrm{~g}, 10 \mathrm{mmol})$ were added successively to a solution of nucleoside ( 9 a ) $(0.367 \mathrm{~g}, 1 \mathrm{mmol})$ in methanol ( 10 ml ) at $-30^{\circ} \mathrm{C}$. The black suspension was stirred for 1 h then acetic anhydride ( 1 ml ) was added. The solution was partitioned between brine $(10 \mathrm{ml})$ and ethyl acetate $(10 \mathrm{ml})$. The two phases were separated and the aqueous layer was extracted twice with ethyl acetate ( 10 ml ). The organic solution was dried $\left(\mathrm{N}_{2} \mathrm{SO}_{4}\right)$ and concentrated under reduced pressure to yield the $4^{\prime}$-acetamido nucleoside (12a) ( $0.167 \mathrm{~g}, 53 \%$ ), m.p. $286{ }^{\circ} \mathrm{C}$ (from dichloromethane-EtOH); $[\alpha]_{\mathrm{D}}^{20}+5^{\circ}(c 0.1$ in MeOH$)$ (Found: C, 54.2; H, 6.7; N, 11.7. $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{6}$ requires C, $54.38 ; \mathrm{H}, 6.56$;
$\mathrm{N}, 11.89 \%$ ); $\lambda_{\text {max. }}(\mathrm{MeOH}) 264 \mathrm{~nm}(\varepsilon 13590) ; \delta_{\mathrm{H}}(300.13 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 1.38$ and $1.62\left(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 1.43(1 \mathrm{H}, \mathrm{dd}, J$ $6.9 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}$ ), 1.94 ( $3 \mathrm{H}, \mathrm{d}, J 1.1 \mathrm{~Hz}, 5-\mathrm{Me}$ ), 2.07 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO}$ ), $4.17\left(1 \mathrm{H}, \mathrm{dd}, J 5\right.$ and $\left.8 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right), 4.39(1 \mathrm{H}, \mathrm{dd}, 6.4$ and 6.9 Hz , $\left.5^{\prime}-\mathrm{H}\right), 4.67\left(1 \mathrm{H}, \mathrm{dd}, J 4.7\right.$ and $\left.5 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 4.75(1 \mathrm{H}$, ddd, $J 4.7$, 6.4 and $\left.8.9 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 5.97\left(1 \mathrm{H}, \mathrm{dd}, J 8 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right), 6.1(1 \mathrm{H}, \mathrm{dd}, J$ $8.9 \mathrm{~Hz}, \mathrm{NH})$, and $7.11(1 \mathrm{H}, \mathrm{dd}, J 1.1,6-\mathrm{H})$.
(E)-7-(4', $6^{\prime}$-Dideoxy-2', $3^{\prime}$-O-isopropylidene-4'-hydroximino-$\alpha$-L-lyxo-hexopyranosyl)theophylline (8b).-Hydroxylamine hydrochloride ( $1.4 \mathrm{~g}, 20 \mathrm{mmol}$ ) was added to a solution of the ketonucleoside ( $\mathbf{7 b}$ ) ( $3.64 \mathrm{~g}, 10 \mathrm{mmol}$ ) in a mixture of ethanol and pyridine ( $1: 1,20 \mathrm{ml}$ ). After being stirred for 1 h , work-up as for compound (3b) afforded the oxime ( 8 b ) $(2.9 \mathrm{~g}, 76 \%)$, m.p. $142{ }^{\circ} \mathrm{C}$ (from EtOH) $[\alpha]_{\mathrm{D}}^{20}-95^{\circ}(c 0.1$ in MeOH ) (Found: C, 48.65; $\mathrm{H}, 5.95 ; \mathrm{N}, 17.55 . \mathrm{C}_{16} \mathrm{H}_{21} \mathrm{~N}_{5} \mathrm{O}_{6} \cdot \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 48.36$; $5.79 ; \mathrm{N}, 17.63 \%) ; \lambda_{\text {max. }}(\mathrm{MeOH}) 275 \mathrm{~nm}(\varepsilon 7996) ; \delta_{\mathbf{H}}(300.13 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 1.42\left(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 1.61$ and $1.63(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}$, $\mathrm{Me}_{2} \mathrm{C}$ and $\left.6^{\prime}-\mathrm{H}\right), 3.42$ and $3.60(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{MeN}), 4.72(1$ $\mathrm{H}, \mathrm{dd}, J 6$ and $\left.6.1 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right), 4.96\left(1 \mathrm{H}, \mathrm{d}, J 6.1 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 5.04(1$ $\left.\mathrm{H}, \mathrm{q}, J 7.1 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 6.41\left(1 \mathrm{H}, \mathrm{d}, J 6.1 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right)$, and $7.42(1 \mathrm{H}$, s, 8-H).
(E)-7-(4'-Acetoximino-4', $6^{\prime}$-dideoxy- $2^{\prime}, 3^{\prime}$-isopropylidene- $\alpha$-L-lyxo-hexopyranosyl)theophylline (9b).-Pyridine ( $0.35 \mathrm{~g}, 4.43$ mmol ), acetic anhydride ( $0.45 \mathrm{~g}, 4.43 \mathrm{mmol}$ ), and a catalytic amount of DMAP were added to a solution of compound (8b) $(1.4 \mathrm{~g}, 3.69 \mathrm{mmol})$ in dichloromethane $(18.5 \mathrm{ml})$. After being stirred for 5 min the reaction mixture concentrated under reduced pressure yielded the title compound ( $\mathbf{9 b}$ ) $(1.242 \mathrm{~g}, 80 \%$ ), m.p. $132-135^{\circ} \mathrm{C}$ (from benzene); $[\alpha]_{\mathrm{D}}^{20}-87.5^{\circ}$ (c 0.1 in MeOH ); (Found: C, 48.6; H, 5.5; N, 16.15. $\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{~N}_{5} \mathrm{O}_{7} \cdot \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 49.20 ; \mathrm{H}, 5.69 ; \mathrm{N}, 15.94 \%$ ); $\lambda_{\text {max. }}$ ( MeOH ) $275 \mathrm{~nm}(\varepsilon$ $7662) ; \delta_{\mathrm{H}}\left(300.13 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.41$ and $1.64(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}$, $\left.\mathrm{Me}_{2} \mathrm{C}\right), 1.68\left(3 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 2.21(3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO}), 3.41$ and $3.60(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{NMe}), 4.74\left(1 \mathrm{H}\right.$, dd, $J 5.8$ and $6.8 \mathrm{~Hz}, 2^{\prime}-$ H), $5.05\left(1 \mathrm{H}, \mathrm{q}, J 7 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 5.11\left(1 \mathrm{H}, \mathrm{d}, J 5.8 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 6.46(1$ $\left.\mathrm{H}, \mathrm{d}, J 6.8 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right)$, and $7.72(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.

7-(4'-Benzyloxycarbonylamino-4', $6^{\prime}$-dideoxy- $2^{\prime}, 3^{\prime}$-O- iso-propylidene- $x-\mathrm{L}-\mathrm{talo}$-hexopyranosyl)theophylline (10b).-Nickel chloride $\left(\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O} ; 0.52 \mathrm{~g}, 2 \mathrm{mmol}\right)$ was added to a solution of the nucleoside ( 9 b ) $(0.42 \mathrm{~g}, 1 \mathrm{mmol})$ in methanol ( 10 ml ) at $-30^{\circ} \mathrm{C}$. To the resulting suspension sodium borohydride $(0.38 \mathrm{~g}$, 10 mmol ) was added portionwise. The reaction mixture was stirred for 30 min , then cooled to $-78^{\circ} \mathrm{C}$, benzyl chloroformate ( $1.195 \mathrm{~g}, 7 \mathrm{mmol}$ ) was added, and the cooled suspension was poured into brine $(10 \mathrm{ml})$ and ethyl acetate ( 10 ml ) immediately. The aqueous phase was extracted with ethyl acetate $(3 \times 10 \mathrm{ml})$ and the combined organic extracts dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporated under reduced pressure. The crude product was purified by flash column chromatography using diethyl ether as eluant to give the carbamate ( 10 b ) $(0.299 \mathrm{~g}, 60 \%$ ), m.p. $96-$ $100^{\circ} \mathrm{C}$ (from diethyl ether and pentane); $[\alpha]_{\mathrm{D}}^{20}-5^{\circ}$ (c 0.1 in MeOH ) (Found: C, $57.9 ; \mathrm{H}, 5.85 ; \mathrm{N}, 13.65 . \mathrm{C}_{24} \mathrm{H}_{29} \mathrm{~N}_{5} \mathrm{O}_{7}$ requires $\mathrm{C}, 57.71 ; \mathrm{H}, 5.81 ; \mathrm{N}, 14.02 \%$ ); $\lambda_{\text {max }}(\mathrm{MeOH}) 275 \mathrm{~nm}(\varepsilon$ 10528 ); $\delta_{\mathrm{H}}\left[300.13 \mathrm{MHz},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right] 1.44$ ( $3 \mathrm{H}, \mathrm{d}, J 6.7 \mathrm{~Hz}, 6^{\prime}-$ $\mathrm{H}), 1.31$ and $1.53\left(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 3.29$ and $3.50(2 \times 3$ $\mathrm{H}, 2 \times \mathrm{s}, \mathrm{NMe}), 4.31\left(1 \mathrm{H}, \mathrm{dq}, J 5\right.$ and $\left.6.7 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 4.36(1 \mathrm{H}$, br dd, $J 5$ and $\left.5.1 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 4.40\left(1 \mathrm{H}, \mathrm{dd}, J 5.1\right.$ and $\left.5.4 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right)$, $4.73\left(1 \mathrm{H}\right.$, dd, $J 5.4$ and $\left.7.8 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right), 5.11(2 \mathrm{H}, \mathrm{s}, 1 \mathrm{H}$, $\left.\mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}\right), 6.21(1 \mathrm{H}, \mathrm{m}, \mathrm{NH}), 6.23\left(1 \mathrm{H}, \mathrm{d}, J 7.8 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right)$, 7.24-7.42 ( $5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}$ ), and $7.42(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.

7-(4'-Amino-4', $6^{\prime}$-dideoxy- $2^{\prime}, 3^{\prime}$ - O-isopropylidene- $\alpha$-L-talohexopyranosyl)theophylline (11b).-(a) From benzyl carbamate (10b). A 125 ml high pressure reactor was charged with compound ( 10 b ) ( $0.499 \mathrm{~g}, 1 \mathrm{mmol}$ ), methanol ( 20 ml ) and $\mathrm{Pd}-\mathrm{C}$
catalyst ( 0.05 g ). The reactor was pressurized with hydrogen at 3.4 atm . After 1 h the suspension was filtered off. The solvent was evaporated off and the amino nucleoside (11b) ( $0.248 \mathrm{~g}, 80 \%$ ) was crystallized.
(b) From oxime (9b). Molybdenum oxide ( $0.36 \mathrm{~g}, 2.5 \mathrm{mmol}$ ) was added to a solution of the nucleoside ( 9 b$)(0.42 \mathrm{~g}, 1 \mathrm{mmol})$ in $\mathrm{MeOH}(10 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$. Sodium borohydride $(0.76 \mathrm{~g}, 20 \mathrm{mmol})$ was added portionwise and the reaction mixture was stirred for 1 h . It was then poured into brine ( 10 ml ) and ethyl acetate ( 10 $\mathrm{ml})$ and the aqueous solution was extracted with ethyl acetate $(3 \times 10 \mathrm{ml})$. The combined organic extracts were dried ( $\mathrm{Na}_{2}{ }^{-}$ $\mathrm{SO}_{4}$ ) then evaporated under reduced pressure. Column chromatography on silica gel with ethyl acetate-methanol ( $80: 20$ ) afforded the $4^{\prime}$-amino nucleoside ( 11 b ) $(0.219 \mathrm{~g}, 60 \%$ ), m.p. $95-$ $98^{\circ} \mathrm{C}$ (from pentane); $[\alpha]_{\mathrm{D}}^{20}-30^{\circ}(c 0.1$ in MeOH ) (Found: C , 51.3; $\mathrm{H}, 6.45 ; \mathrm{N}, 18.2 . \mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{5} \mathrm{O}_{5} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ requires C , 51.33 ; $\mathrm{H}, 6.41 ; \mathrm{N}, 18.71 \%$ ); $\lambda_{\text {max. }} .(\mathrm{MeOH}) 275 \mathrm{~nm}(\varepsilon 8650) ; \delta_{\mathrm{H}}(300.13$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.45\left(3 \mathrm{H}, \mathrm{d}, J 6.7 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 1.36$ and $1.58(2 \times 3$ $\left.\mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 3.41$ and $3.6(3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{MeN}), 3.53(1 \mathrm{H}, \mathrm{dq}, J$ 3.6 and $\left.6.7 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 4.32\left(1 \mathrm{H}, \mathrm{dd}, J 3.6\right.$ and $\left.5.3 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 4.60$ $\left(1 \mathrm{H}, \mathrm{dd}, J 5.3\right.$ and $\left.6.8 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 4.76(1 \mathrm{H}, \mathrm{dd}, J 6.8$ and 8.1 Hz , $\left.2^{\prime}-\mathrm{H}\right), 6.14(1 \mathrm{H}, \mathrm{d}, J 8.1 \mathrm{~Hz})$, and $7.79(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.

7-(4'-Acetamido-4', $6^{\prime}$-dideoxy- $2^{\prime}, 3^{\prime}$-isopropylidene- $\alpha$-L-talohexopyranosyl)theophylline (12b).-Nickel chloride $\left(\mathrm{NiCl}_{2}{ }^{-}\right.$ $6 \mathrm{H}_{2} \mathrm{O} ; 1.14 \mathrm{~g}, 4 \mathrm{mmol}$ ) and sodium borohydride ( $0.76 \mathrm{~g}, 20$ mmol ) were added successively to a solution of the nucleoside (9b) $(0.842 \mathrm{~g}, 2 \mathrm{mmol})$ in methanol ( 20 ml ) at $-30^{\circ} \mathrm{C}$. The black suspension was stirred for 1 h then acetic anhydride ( 2 ml ) was added. The solution was partitioned between brine $(20 \mathrm{ml})$ and ethyl acetate ( 20 ml ), the two phases were separated, and the aqueous layer was extracted with ethyl acetate $(2 \times 20 \mathrm{ml})$. The organic solution was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent evaporated off under reduced pressure to give the $4^{\prime}$-acetamido nucleoside ( $\mathbf{1 2 b}$ ) ( $0.56 \mathrm{~g}, 70 \%$ ), m.p. $114-115^{\circ} \mathrm{C}$ (from MeOH ); $[x]_{\mathrm{D}}^{2 \mathrm{O}}-10^{\circ}(c 0.1$ in MeOH ) (Found: C, 51.7; H, 6.5; N, 15.9. $\mathrm{C}_{18} \mathrm{H}_{25} \mathrm{~N}_{5} \mathrm{O}_{6} \cdot \mathrm{MeOH}$ requires $\mathrm{C}, 52.04 ; \mathrm{H}, 6.43 ; \mathrm{H}, 15.97 \%$ ); $\lambda_{\text {max. }}(\mathrm{MeOH}) 275 \mathrm{~nm}(\varepsilon 7692) ; \delta_{\mathrm{H}}\left(300.13 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.38$ and $1.56\left(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 1.52\left(3 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 2.08$ $(3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO}), 3.51$ and $3.6(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{NMe}), 4.43(1 \mathrm{H}$, $\mathrm{dq}, J 6.7$ and $\left.7 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 4.53\left(1 \mathrm{H}, \mathrm{dd}, J 5.2\right.$ and $\left.7.5 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right)$, 4.57 ( 1 H, dd, $J 4.4$ and $5.2 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}$ ), 4.67 ( 1 H , ddd, $J 4.4,6.7$, and $\left.8.9 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 5.96(1 \mathrm{H}, \mathrm{d}, J 8.9 \mathrm{~Hz}, \mathrm{NH}), 6.24(1 \mathrm{H}, \mathrm{d}, J 7.5$ $\left.\mathrm{Hz}, 1^{\prime}-\mathrm{H}\right)$, and $7.86(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.
(Z)-7-( $2^{\prime}, 3^{\prime}, 4^{\prime}, 6^{\prime}-$ Tetradeoxy-4'-hydroximino- $\beta$-L-glycero-hex-$2^{\prime}$-enopyranosyl)theophylline (14a).--The keto nucleoside (13) $(1.9 \mathrm{~g}, 10 \mathrm{mmol})$ and hydroxylamine hydrochloride $(2.8 \mathrm{~g}, 40$ $\mathrm{mmol})$ were stirred in a mixture of pyridine and ethanol $(1: 1,20$ $\mathrm{ml})$ for 5 min . The reaction mixture was partitioned between water and brine, the aqueous phase was extracted with ethyl acetate ( $3 \times 25 \mathrm{ml}$ ) and the organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. Evaporation of the solvent yielded the nucleoside (14a) $\left(2.44 \mathrm{~g}, 80 \%\right.$ ), m.p. $238-239^{\circ} \mathrm{C}$ (from MeOH ); $[\alpha]_{\mathrm{D}}^{20}$ $+100^{\circ}\left(c 0.1\right.$ in $\left.\mathrm{CH}_{3} \mathrm{COOH}\right)$ (Found: C, 51.3; H, 6.45; N, 18.2. $\mathrm{C}_{16} \mathrm{H}_{23} \mathrm{~N}_{5} \mathrm{O}_{5} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ requires C, $51.33 ; \mathrm{H}, 6.41 ; \mathrm{N}, 18.71 \%$ ); $\delta_{\mathrm{H}}\left(60 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.6\left(3 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 3.45$ and 3.6 $(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{MeN}), 4.8\left(1 \mathrm{H}, \mathrm{q}, J 6.5 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 6.5(1 \mathrm{H}, \mathrm{dd}$, $J 1.5$ and $\left.10 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 7.25\left(1 \mathrm{H}\right.$, br s, $\left.1^{\prime}-\mathrm{H}\right), 7.5(1 \mathrm{H}, \mathrm{dd}, J 1.8$ and $\left.10 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right)$, and $8.2(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.
(Z)-7-(4'-Acetoximino-2', $3^{\prime}, 4^{\prime}, 6^{\prime}$-tetradeoxy- $\beta$-L-glycero-hex-$2^{\prime}$-enopyranosyl)theophylline (14b).-The nucleoside (14a) (1.4 $\mathrm{g}, 4.6 \mathrm{mmol}$ ) in dichloromethane ( 20 ml ) was treated with acetic anhydride ( $2.04 \mathrm{~g}, 20 \mathrm{mmol}$ ) in the presence of pyridine $(1.58 \mathrm{~g}$, 20 mmol ) and a catalytic amount of 4-dimethylaminopyridine to give after 1 h the title compound ( $\mathbf{1 4 b}$ ) $(1.276 \mathrm{~g}, 80 \%)$, m.p. $155-156^{\circ} \mathrm{C}$ (from EtOH); $[\alpha]_{\mathrm{D}}^{20}+112^{\circ}(c 0.1$ in MeOH$)$
(Found: C, $51.95 ; \mathrm{H}, 4.95$; N, 20.2. $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{O}_{5}$ requires C, $51.8 ; \mathrm{H}, 4.89 ; \mathrm{N}, 20.17 \%) ; \lambda_{\text {max. }}(\mathrm{MeOH}) 273 \mathrm{~nm}(\varepsilon 10965) ; \delta_{\mathrm{H}}(60$ $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $1.6\left(3 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 2.05(3 \mathrm{H}, \mathrm{s}, \mathrm{MeCO})$, 3.45 and $3.6(3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{MeN}), 4.7\left(1 \mathrm{H}, \mathrm{q}, 6.5 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 6.5(1$ $\mathrm{H}, \mathrm{dd}, J 1.5$ and $\left.10 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 7.05\left(1 \mathrm{H}\right.$, br s, $\left.1^{\prime}-\mathrm{H}\right), 7.2(1 \mathrm{H}, \mathrm{dd}, J$ 1.8 and $\left.10 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}\right)$, and $7.8(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.

7-(4'-Benzyloxycarbonylamino- $2^{\prime}, 3^{\prime}, 4^{\prime}, 6^{\prime}$-tetradeoxy- $\beta$-L-erythro-hexopyranosyl)theophylline (15a) and 7-(4'-Benzyloxy-carbonylamino- $2^{\prime}, 3^{\prime}, 4^{\prime}, 6^{\prime}$-tetradeoxy- $\beta$-L-threo-hexopyranosyl)theophylline (15b).—Nickel chloride $\left(\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O} ; 1.04 \mathrm{~g}, 4\right.$ $\mathrm{mmol})$ was added to a solution of the nucleoside (14b) $(0.694 \mathrm{~g}$, 2 mmol ) in methanol ( 20 ml ) at $-15^{\circ} \mathrm{C}$ followed by sodium borohydride ( $0.76 \mathrm{~g}, 20 \mathrm{mmol}$ ) portionwise. After being stirred for 30 min the reaction mixture was cooled to $-78^{\circ} \mathrm{C}$, benzyl chloroformate ( $1.195 \mathrm{~g}, 7 \mathrm{mmol}$ ) was added, and the cooled suspension was poured into brine ( 10 ml ) and ethyl acetate ( 10 ml ) immediately. The aqueous phase was extracted with ethyl acetate ( $3 \times 10 \mathrm{ml}$ ) and the combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. Evaporation of the solvent under reduced pressure and flash column chromatography using first diethyl ether then ethyl acetate afforded the benzyloxycarbonylamino nucleosides ( 15 a ) $(0.213 \mathrm{~g}, 25 \%)$ and ( $\mathbf{1 5 b}$ ) $(0.213 \mathrm{~g}, 25 \%$ ).

Compound (15a) had m.p. $205^{\circ} \mathrm{C}$ (from EtOH); $[\alpha]_{\mathrm{D}}^{2 \mathrm{O}}-20^{\circ}$ (c 0.1 in $\mathrm{CHCl}_{3}$ ) (Found: C, 58.8; H, 5.95; N, 16.1. $\mathrm{C}_{21} \mathrm{H}_{25} \mathrm{~N}_{5} \mathrm{O}_{5}$ requires $\mathrm{C}, 59.01 ; \mathrm{H}, 5.85 ; \mathrm{N}, 16.39 \%$ ); $\lambda_{\text {max. }}$. $(\mathrm{MeOH}) 277 \mathrm{~nm}(\varepsilon$ $12126)$; $\delta_{\mathrm{H}}\left(300.13 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.23\left(3 \mathrm{H}, \mathrm{d}, J 6.1 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right)$, $1.81\left(1 \mathrm{H}\right.$, dddd, $J 5.11,12.23,12.26$, and $\left.12.52 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}_{\mathrm{a}}\right), 2.12-$ $2.24\left(\mathrm{~m}, 3 \mathrm{H}, 2^{\prime}-\mathrm{H}_{\mathrm{e}}, 2^{\prime}-\mathrm{H}_{\mathrm{a}}\right.$, and $\left.3^{\prime}-\mathrm{H}_{\mathrm{e}}\right), 3.29$ and $3.49(2 \times 3 \mathrm{H}$, $2 \times \mathrm{s}, \mathrm{MeN}), 3.47\left(1 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}\right), 3.70(1 \mathrm{H}, \mathrm{dd}, J 6.1$ and 9.89 $\left.\mathrm{Hz}, 5^{\prime}-\mathrm{H}\right), 5.09\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}\right), 6.02(1 \mathrm{H}, \mathrm{dd}, J 3.4$ and 9.6 $\left.\mathrm{Hz}, 1^{\prime}-\mathrm{H}\right), 6.44(1 \mathrm{H}, \mathrm{d}, J 9 \mathrm{~Hz}, \mathrm{NH}), 7.25-7.40\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right)$, and $8.18(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.

Compound (15b) had m.p. $190^{\circ} \mathrm{C}$ (from EtOH); $[\alpha]_{\mathrm{D}}^{20} 0^{\circ}(c$ 0.1 in $\mathrm{CHCl}_{3}$ ) (Found: C, $58.85 ; \mathrm{H}, 5.75 ; \mathrm{N}, 16.35$ ); $\lambda_{\text {max }}$. $(\mathrm{MeOH})$ $277 \mathrm{~nm}(\varepsilon 11827)$; $\delta_{\mathbf{H}}\left(300.13 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.13(3 \mathrm{H}, \mathrm{d}, J 6.4$ $\left.\mathrm{Hz}, 6^{\prime}-\mathrm{H}\right), 1.90-2.04\left(3 \mathrm{H}, \mathrm{m}, 2^{\prime}-\mathrm{H}_{\mathrm{e}}, 3^{\prime}-\mathrm{H}_{\mathrm{a}}\right.$, and $3^{\prime}-\mathrm{H}_{\mathrm{e}}$ ), 2.48 (1 H , dddd, $J 6.5,11.2,12.13$, and $12.13 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}_{\mathrm{a}}$ ), 3.33 and 3.5 $(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{MeN}), 3.81\left(1 \mathrm{H}\right.$, br d, $\left.J 8.7 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 4.01(1$ H , dd, $J 1.6$ and $\left.6.4 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 5.13\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}\right), 5.95(1 \mathrm{H}$, dd, $J 3$ and $\left.11.2 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right), 7.09(1 \mathrm{H}, \mathrm{d}, J 8.7 \mathrm{~Hz}, \mathrm{NH}), 7.09-7.27$ $\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right)$, and $8.16(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.

Reduction of $(\mathrm{Z})$-7-(4'-Acetoximino- $2^{\prime}, 3^{\prime}, 4^{\prime}, 6^{\prime}$-tetradeoxy- $\beta$-L-glycero-hex-2'-enopyranosyl)theophylline (14b) in the Presence of $\mathrm{MoO}_{3}$.-Molybdenum(vi) oxide ( $0.202 \mathrm{~g}, 1.4 \mathrm{mmol}$ ) was added to a solution of nucleoside (14b) $(0.305 \mathrm{~g}, 1 \mathrm{mmol})$ in methanol ( 10 ml ). After being cooled to $0^{\circ} \mathrm{C}$, the blue suspension was treated with sodium borohydride $(0.76 \mathrm{~g}, 20$ mmol ). After 30 min the reaction mixture was treated in the following manner.
(a) The suspension was poured into brine $(10 \mathrm{ml})$ and diethyl ether $(10 \mathrm{ml})$ and the aqueous phase was extracted with diethyl ether ( $3 \times 10 \mathrm{ml}$ ). The combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the diethyl ether evaporated under reduced pressure. Column chromatography on silica gel with dichloro-methane-acetone (3:7) afforded the $4^{\prime}$-amino nucleosides (16a) $(0.106 \mathrm{~g}, 25 \%)$ and (17) ( $0.106 \mathrm{~g}, 25 \%$ ).

Compound (16a) had m.p. $182-183^{\circ} \mathrm{C}$ (from acetone); $[\alpha]_{\mathrm{D}}^{20}$ $-170^{\circ}$ (c $0.1 \mathrm{in} \mathrm{MeOH);} \lambda_{\text {max. }}$. (MeOH) 275 nm ( $\varepsilon 9544$ ); $\delta_{\mathrm{H}}\left(300.13 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.39\left(3 \mathrm{H}, \mathrm{d}, J 6.1 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 3.34(1 \mathrm{H}$, ddd, $J 1.8,2.1$, and $\left.8.36 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right)$, 3.42 and $3.60(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}$, $\mathrm{Me}-\mathrm{N}), 3.71\left(1 \mathrm{H}, \mathrm{dq}, J 6.1\right.$ and $\left.8.36 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 5.92(1 \mathrm{H}$, ddd, $J$ $1.8,2.1$, and $\left.10 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 6.15\left(1 \mathrm{H}\right.$, ddd, $J 1.8,2.1$, and $10 \mathrm{~Hz}, 2^{\prime}-$ H), $6.78\left(1 \mathrm{H}, \mathrm{dd}, J 1.8\right.$ and $\left.2.1 \mathrm{~Hz}, 1^{\prime}-\mathrm{H}\right)$, and $7.78(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.

Compound (17) had $\delta_{\mathrm{H}}\left(300.13 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.34(3 \mathrm{H}, \mathrm{d}, J$ $\left.6.1 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 1.58(1 \mathrm{H}$, dddd, $J 3.6,11.17,12.95$, and 13 Hz , $\left.3^{\prime}-\mathrm{H}_{\mathrm{a}}\right), 1.84\left(1 \mathrm{H}\right.$, dddd, $J 3.96,10.8,12.95$, and $\left.13.06 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}_{\mathrm{a}}\right)$,
2.12 ( 1 H , dddd, $J 3.6,3.96,4.05$, and $13 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}_{\mathrm{e}}$ ), $2.30(1 \mathrm{H}$, dddd, $J 2.2,3.6,3.6$, and $\left.13.06 \mathrm{~Hz}, 2^{\prime}-\mathrm{H}_{\mathrm{e}}\right), 2.58(1 \mathrm{H}$, ddd, $J 4.05$, 9.25 , and $11.17 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}$ ), 3.41 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}$ ), 3.46 ( 1 H , overlapped, $\left.5^{\prime}-\mathrm{H}\right)$, $3.6(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 5.98(1 \mathrm{H}, \mathrm{dd}, J 2.2$ and 10.8 $\left.\mathrm{Hz}, 1^{\prime}-\mathrm{H}\right)$, and $7.86(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.
(b) The black suspension was stirred for 30 min then benzyl chloroformate was added. The solution was partitioned between a mixture of brine $(10 \mathrm{ml})$ and ethyl acetate $(10 \mathrm{ml})$. The aqueous phase was extracted with ethyl acetate ( $3 \times 10 \mathrm{ml}$ ). The organic solution was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated under reduced pressure. The resulting yellow syrup was chromatographed on silica gel (flash chromatography) with diethyl ether then ethyl acetate to give compounds (16b) (0.106 $\mathrm{g}, 25 \%$ ) and ( 15 a ) ( $0.106 \mathrm{~g}, 25 \%$ ).

Compound (16b) had m.p. $198^{\circ} \mathrm{C}$ (from EtOH); $[\alpha]_{\mathrm{D}}^{20}-75^{\circ}$ (c 0.1 in $\mathrm{CDCl}_{3}$ ) (Found $\mathrm{C}, 58.4 ; \mathrm{H}, 5.4 ; \mathrm{N}, 15.9$. $\mathrm{C}_{21} \mathrm{H}_{25} \mathrm{~N}_{5} \mathrm{O}_{5} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 58.06 ; \mathrm{H}, 5.52 ; \mathrm{N}, 16.12 \%$; ; $\lambda_{\text {max. }}(\mathrm{MeOH}) 276 \mathrm{~nm}(\varepsilon 9223) ; \delta_{\mathrm{H}}\left(300.13 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.29(3$ $\left.\mathrm{H}, \mathrm{d}, J 6.2 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 3.29$ and $3.5(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{MeN}), 3.91(1$ H , dd, $J 6.2$ and $\left.9.04 \mathrm{~Hz}, 5^{\prime}-\mathrm{H}\right), 4.23(1 \mathrm{H}$, dddd, $J 1.5,1.7,8.7$, and $\left.9.04 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 5.12\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2} \mathrm{C}_{6} \mathrm{H}_{5}\right), 6.03(1 \mathrm{H}$, ddd, $J 1.5$, 1.7 , and $\left.10.3 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 6.11\left(1 \mathrm{H}\right.$, ddd, $J 1.5,2.3$, and $10.3 \mathrm{~Hz}, 2^{\prime}-$ H), $6.66(1 \mathrm{H}, \mathrm{d}, J 8.7 \mathrm{~Hz}, \mathrm{NH}), 6.76\left(1 \mathrm{H}, \mathrm{dd}, J 1.5\right.$ and $2.3 \mathrm{~Hz}, 1^{\prime}-$ H), $7.27-7.41\left(5 \mathrm{H}, \mathrm{m}, \mathrm{C}_{6} \mathrm{H}_{5}\right)$, and $8.08(1 \mathrm{H}, \mathrm{s}, 8-\mathrm{H})$.

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[^0]:    * Despite various experiments the nature of the interaction revealed by the n.m.r. spectrum could not be clearly established. The n.m.r. spectrum of compound (5a) exhibited a temperature and solvent dependency. In $\mathrm{CDCl}_{3}$ variation of the temperature ( $25-55^{\circ} \mathrm{C}$ ) had a marked effect upon the equilibria affording exclusively the minor component of the mixture. In $\left[{ }^{2} \mathrm{H}_{6}\right.$ ]acetone this transformation occurred in 5 min at $25^{\circ} \mathrm{C}$ whereas in deuteriated dimethyl sulphoxide heating above $120^{\circ} \mathrm{C}$ for several hours was required. Finally after cooling of the observed samples no formation of the major component of the mixture could be detected even after several days.

[^1]:    * For (3a) $\delta_{\mathbf{H}}\left(60 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.35\left(3 \mathrm{H}, \mathrm{d}, J 6.5 \mathrm{~Hz}, 6^{\prime}-\mathrm{H}\right), 1.4$ and 1.57 $\left(2 \times 3 \mathrm{H}, 2 \times \mathrm{s}, \mathrm{Me}_{2} \mathrm{C}\right), 1.8(3 \mathrm{H}, \mathrm{s}, 5-\mathrm{Me}), 3.7(1 \mathrm{H}, \mathrm{dq}, J 2$ and 6.5 Hz , $\left.5^{\prime}-\mathrm{H}\right), 4.3\left(1 \mathrm{H}, \mathrm{dd}, J 2\right.$ and $\left.7 \mathrm{~Hz}, 4^{\prime}-\mathrm{H}\right), 5\left(0.66 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 5.9$ $\left(0.33 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 3^{\prime}-\mathrm{H}\right), 6.5\left(0.33 \mathrm{H}, \mathrm{s}, 1^{\prime}-\mathrm{H}\right), 6.7\left(0.66 \mathrm{H}, \mathrm{s}, 1^{\prime}-\mathrm{H}\right), 7.7$ $(0.33 \mathrm{H}, \mathrm{s}, 6-\mathrm{H})$, and $7.9(0.66 \mathrm{H}, \mathrm{s}, 6-\mathrm{H})$.

